

PATENT NUMBER

APR 10 1968 I.P.E.
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PATENT DATE

TITLE	APPLICANTS
1. The effect of the concentration of the solution on the rate of reaction.	Mr. J. H. Smith
2. The effect of the temperature of the solution on the rate of reaction.	Mr. J. H. Smith
3. The effect of the surface area of the solid reactant on the rate of reaction.	Mr. J. H. Smith
4. The effect of the catalyst on the rate of reaction.	Mr. J. H. Smith
5. The effect of the concentration of the solution on the equilibrium constant.	Mr. J. H. Smith
6. The effect of the temperature of the solution on the equilibrium constant.	Mr. J. H. Smith
7. The effect of the surface area of the solid reactant on the equilibrium constant.	Mr. J. H. Smith
8. The effect of the catalyst on the equilibrium constant.	Mr. J. H. Smith
9. The effect of the concentration of the solution on the rate of reaction.	Mr. J. H. Smith
10. The effect of the temperature of the solution on the rate of reaction.	Mr. J. H. Smith
11. The effect of the surface area of the solid reactant on the rate of reaction.	Mr. J. H. Smith
12. The effect of the catalyst on the rate of reaction.	Mr. J. H. Smith
13. The effect of the concentration of the solution on the equilibrium constant.	Mr. J. H. Smith
14. The effect of the temperature of the solution on the equilibrium constant.	Mr. J. H. Smith
15. The effect of the surface area of the solid reactant on the equilibrium constant.	Mr. J. H. Smith
16. The effect of the catalyst on the equilibrium constant.	Mr. J. H. Smith
17. The effect of the concentration of the solution on the rate of reaction.	Mr. J. H. Smith
18. The effect of the temperature of the solution on the rate of reaction.	Mr. J. H. Smith
19. The effect of the surface area of the solid reactant on the rate of reaction.	Mr. J. H. Smith
20. The effect of the catalyst on the rate of reaction.	Mr. J. H. Smith

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☐ The term of this patent shall _____ (date) not extend beyond the expiration date of U.S. Patent No. _____

☐ The term of _____ months of this patent have been disclaimed.

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